This article was downloaded by: [University of Haifa Library]

On: 11 August 2012, At: 10:57 Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered

office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and subscription information:

http://www.tandfonline.com/loi/gmcl20

# g -Tensor Analyses of $\beta'$ -Type Pd(dmit) 2 Metal Complexes

Toshikazu Nakamura <sup>a</sup> , Toshihiro Takahashi <sup>b</sup> , Shuji Aonuma <sup>c</sup> & Reizo Kato <sup>d</sup>

<sup>a</sup> Institute for Molecular Science, Okazaki, 444-8585, Japan

<sup>b</sup> Department of Physics, Gakushuin University, Mejiro 1-5-1, Toshimaku, Tokyo, 171-8588, Japan

<sup>c</sup> Department of Materials Science, Osaka Electro-Communication University, 18-8 Hatsu-Cho, Neyagawa, Osaka, 572-8530, Japan

<sup>d</sup> The Institute of Physics and Chemical Research, 2-1 Hirosawa, Wako-shi, Saitama, 351-0198, Japan

Version of record first published: 18 Oct 2010

To cite this article: Toshikazu Nakamura, Toshihiro Takahashi, Shuji Aonuma & Reizo Kato (2003): g -Tensor Analyses of  $\beta'$ -Type Pd(dmit) 2 Metal Complexes, Molecular Crystals and Liquid Crystals, 379:1, 53-58

To link to this article: <a href="http://dx.doi.org/10.1080/713738621">http://dx.doi.org/10.1080/713738621</a>

#### PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <a href="http://www.tandfonline.com/page/terms-and-conditions">http://www.tandfonline.com/page/terms-and-conditions</a>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

*Mol. Cryst. Liq. Cryst.*, Vol. 379, pp. 53-58 Copyright © 2002 Taylor & Francis 1058-725X/02 \$12.00 + .00 DOI: 10.1080/10587250290090255



### g-Tensor Analyses of β'-Type Pd(dmit)<sub>2</sub> Metal Complexes

# TOSHIKAZU NAKAMURA<sup>a</sup>, TOSHIHIRO TAKAHASHI<sup>b</sup>, SHUJI AONUMA<sup>c</sup> and REIZO KATO<sup>d</sup>

<sup>a</sup>Institute for Molecular Science, Okazaki 444-8585, Japan,
<sup>b</sup>Department of Physics, Gakushuin University, Mejiro 1-5-1, Toshimaku,
Tokyo 171-8588, Japan,
<sup>c</sup>Department of Materials Science, Osaka Electro-Communication University,
18-8 Hatsu-Cho, Neyagawa, Osaka 572-8530, Japan and
<sup>d</sup>The Institute of Physics and Chemical Research, 2-1 Hirosawa, Wako-shi,
Saitama, 351-0198, Japan

ESR measurements and g-tensor analyses were performed for metal complexes,  $\beta$ '-type  $Pd(dmit)_2$ . The ESR g-values of  $\beta$ '-type  $Pd(dmit)_2$  are found to be beyond one radical description which is a good approximation for conventional molecular based conductors. In this paper we focus on the anomalous behavior of the ESR g-values in  $\beta$ '-type  $Pd(dmit)_2$ . We discuss the possible explanation of the electronic structure of  $\beta$ '-type  $Pd(dmit)_2$  metal complexes from microscopic points of view.

Keyword metal complexes; dmit; ESR

#### INTRODUCTION

Metal dithiolene complexes,  $Pd(dmit)_2$  (where the dmit is the 1,3-dithia-2-thione-4,5-dithiolato), are one of the major families in organic conductors. While  $R_4Z[Pd(dmit)_2]_2$  compounds show metallic behavior at room temperature, the low temperature phases are

paramagnetic insulators.<sup>1,2</sup> In order to clarify the mechanism of the charge localization of  $\beta$ '-type Pd(dmit)<sub>2</sub> compounds, we carried out magnetic resonance measurements.<sup>3-7</sup> Although they are isostrcural and with little differences in lattice parameters, the ESR parameters show different behavior with counter cations. As mentioned in the previous report,<sup>7</sup> we found a close relation between the antiferromagnetic transition temperature,  $T_{\rm N}$  and inter-stack interaction. However, several problems remain unsolved. In this paper, we focus on the anomalous behavior of the ESR g-values in  $\beta$ '-type Pd(dmit)<sub>2</sub>. We discuss the electronic structure of  $\beta$ '-type Pd(dmit)<sub>2</sub> metal complexes from microscopic points of view.

#### **EXPERIMENTAL**

Sample preparation and crystal structural data were shown in previous reports.  $^{2,8-10}$  The ESR measurements were carried out for  $1\sim 10$  aligned single crystal(s) using an X-band spectrometer JES-FE3XG (JEOL) equipped with an Air Products continuous-flow cryostat LTR-3 (between 100 K and 7 K); experimental details were shown in ref. 11. In order to investigate lower temperatures, the Et<sub>2</sub>Me<sub>2</sub>Sb was also measured using an X-band spectrometer ESP-300E (Bruker) equipped with an Oxford continuous-flow cryostat down to 4 K. We confirmed the consistent of the results measured by both equipments.

#### RESULTS AND DISCUSSION

The ESR measurements were performed for  $\beta$ '- $R_4Z[Pd(dmit)_2]_2$  with counter cations of Me<sub>4</sub>Z (Z= As and Sb) and Et<sub>2</sub>Me<sub>2</sub>Z (Z= P, As and Sb). Hereafter, we classify the experimental results into following three groups; (a) compounds with  $T_N$  around 35 K, (b) compounds with  $T_N$  around 18 K, and (c) compounds with  $T_N$  below experimental limitation (4 K). Since no significant difference was found among

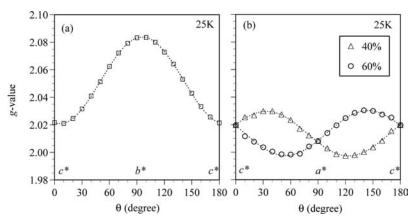


FIGURE 1 Angular dependence of the g-values of of  $Et_2Me_2P$  salts at 25 K.

the compounds in the same group, we show the experimental results for the typical compounds of each groups; the results of (a) Me<sub>4</sub>As, (b)Et<sub>2</sub>Me<sub>2</sub>P and (c)Et<sub>2</sub>Me<sub>2</sub>Sb cations are shown. Scattering of data at high temperatures is due to weak ESR signals associated with huge broadening of the linewidth.

The anisotropy of the g-values is huge even in the case of the external field applying in the conduction plane. These features are different from TTF based organic conductors where the in-plane anisotropy of the g-values is small and the g-value shows a maximum when the external field applied perpendicular to the conducting plane. Angular dependence of the g-values of  $Et_2Me_2P$  at 25 K was shown in Fig. 1. While the resonance signal is single in the  $a^*b^*$  plane rotation, it splits into two within the  $c^*a^*$  plane. The ratio of the relative intensity is approximately 6:4. Since the crystal structure is monoclinic and we used 10 aligned crystals for this measurement, two arrangements are possible when we put the  $a^*b^*$  plane of the crystals parallel to the ESR sample stage. Considering

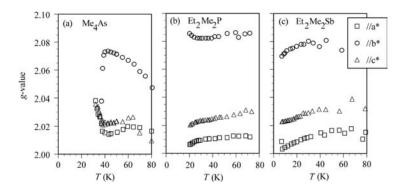


FIGURE 2 Temperature dependence of the g-values of  $\beta$ '-type  $Pd(dmit)_2$  compounds; (a)  $Me_4As$ , (b)  $Et_2Me_2P$  and (c)  $Et_2Me_2Sb$  cations  $(H_0||a^*(square), ||b^*(circle), ||c^*(triangle))$ .

this situation, we can evaluate the principal values and axes of the g-tensor by least squares fit to the measured data assuming anisotropic g-value.

The principal axes are well explained as the average of the molecular axes of  $Pd(dmit)_2$  radicals with solid crossing columns. On the other hand, the principal values of  $Pd(dmit)_2$  show characteristic features. The large shifts from  $g_c$  of the  $g_1$  and  $g_2$  are remarkable, while the small shift of the  $g_3$  is explained by the fact that HOMO and/or LUMO are mainly composed of orbital elongating the z direction. Considering the huge shift of the  $g_1$ , it seems likely that the Pd  $d_{3z^2-r^2}$  orbital also composes of the wave function besides the  $p_z$  orbital of C and S. It is noted that the principal value shows maximum for approximately molecular short axis, while the maximum principal values of TTF derivatives are along the molecular long axes. This fact suggests that there is small contribution of the Pd d orbital along the molecular long axis. However, an isolated

Pd(dmit)<sub>2</sub> possesses a HOMO without the d contribution because of its symmetry according to the extended Hückel Molecular orbital calculation. Moreover, the average of the principal values,  $\langle g_{\rm AV} \rangle$  (= 2.038), of  $\beta$ '-Et<sub>2</sub>Me<sub>2</sub>P salt does not agree with that of a 1:1 radical, Bu<sub>4</sub>N[Pd(dmit)<sub>2</sub>] (Bu=butyl) solution,  $\langle g \rangle$  (= 2.0191). The observed g-tensor cannot be explained by that of the isolated Pd(dmit)<sub>2</sub> molecule. It is likely that the discrepancy of the g-tensor from the isolated radical is due to the deformation of the wave-function with the strong dimer structure of the Pd(dmit)<sub>2</sub> system. Recent first-principle calculations by Miyazaki et~al. including the actual dimer effect suggest that the HOMO of Pd(dmit)<sub>2</sub> has appreciable contributions from the Pd  $d_{3z^2-r^2}$  orbitals. This expectation is coincided with the ESR results.

Figure 2 shows the temperature dependence of the g-values of typical compounds of three groups; (a) Me<sub>4</sub>As, (b)Et<sub>2</sub>Me<sub>2</sub>P and (c)Et<sub>2</sub>Me<sub>2</sub>Sb cations, applying the external static field along the crystal axes. The g-values for all the directions are changing their absolute values with temperature. The change is pronounced in the  $b^*$ axis; approximately parallel to the molecular short axis. Especially, Me<sub>4</sub>As salt shows crucial temperature dependence. Since the system possesses one kind of spin species, we cannot explain the anomalous behavior of the q-value by the orientation of the molecules. One of possible explanations is antiferromagnetic fluctuation. Significant gshifts are observed in one-dimensional AF magnets just above  $T_N$ 's. However, in the case of Pd(dmit)<sub>2</sub>, the onset temperatures of the g-shift seem to be too high considering the  $T_{\rm N}$ . Antiferromagnetic fluctuation seems unlikely. Another possible explanation is a dimerization effect. As discussed above, the discrepancy of the principal values from those of the isolated radical is considered to be due to the strong dimer structure of the  $Pd(dmit)_2$  system. Hence it seems likely that possible change of the magnitude of the dimerization causes the curious temperature dependence of the g-values. Further investigations are now going on.

In conclusion, we measured the g-values and performed the g-tensor analyses for the family of  $Pd(dmit)_2$  metal complexes. The ESR g-values of  $Pd(dmit)_2$  are found to be beyond one radical description; we should consider the wave-function  $[Pd(dmit)_2]_2$  supramolecules.

#### Acknowledgments

This work was supported by the Grant-in-Aid for Scientific Research on Priority Area "Metal-assembled Complexes" (No. 10149245) from the Ministry of Education, Science, Sports and Culture of Japan.

#### References

- [1] A. Kobayashi, H. Kim, Y. Sasaki, K. Murata, R. Kato and H. Kobayashi, J. Chem. Soc. Faraday Trans., 86, 361 (1990).
- [2] R. Kato, Y.-L. Liu, S. Aonuma and H. Sawa, <u>Solid State Commun.</u> 98, 1021 (1995).
- [3] K. Seya, Y. Kobayashi, T. Nakamura, T. Takahashi, Y. Osako, H. Kobayashi, R. Kato, A. Kobayashi and H. Iguchi, Synth. Met., 70, 1043 (1995).
- [4] Y. Tsuchiya, T. Nakamura, T. Takahashi, Y.-L. Liu, H. Sawa and R. Kato, Synth. Met., **86**, 2117 (1997).
- [5] T. Nakamura, H. Yamane, T. Takahashi, S. Aonuma and R. Kato, Synth. Met., 103, 2142 (1999).
- [6] T. Nakamura, H. Tsukada, T. Takahashi, S. Aonuma and R. Kato, Mol. Cryst. Liq. Cryst., 343, 187 (2000).
- [7] T. Nakamura, T. Takahashi, S. Aonuma and R. Kato,
- J. Mater. Chem., 11 2195 (2001).
- [8] S. Aonuma, H. Sawa and R. Kato, Synth. Met., 86, 1881 (1997).
- [9] R. Kato, Y.-L. Liu, Y. Hosokoshi and S. Aonuma,
- Mol. Cryst. Liq. Cryst., 296, 217 (1997).
- [10] S. Rouziére, J.-I. Yamaura and R. Kato, <u>Phys. Rev.</u>, **B60**, 3113 (1999).
- [11] T. Nakamura, T. Nobutoki, T. Takahashi, G. Saito, H. Mori and T. Mori, J. Phys. Soc. Jpn., 63, 4110 (1994).
- [12] T. Miyazaki and T. Ohno, Phys. Rev., **B59**, R5269 (1999).